

Activity On Ionic Bonding With Answers

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chemical compound composed of ions held together by electrostatic forces termed ionic bonding the compound is neutral overall but consists of positively charged ions called cations and negatively charged ions called anions these can be simple ions such as the sodium na and chloride cl in sodium chloride or

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electronegative element is assigned a value of 4.0 and values range down to caesium and francium which are the least electronegative at 0.7
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carbon bonding the carbon atom is unique among elements in its
tendency to form extensive networks of covalent bonds not only with
other elements but also with itself because of its position midway in the
second horizontal row of the periodic table carbon is neither an
electropositive nor an electronegative element it therefore is more likely
to

which of the following molecules may have hydrogen bonds Sep 15
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can have hydrogen bonds between other molecules of the same kind to
have hydrogen bonding you need an N O or F atom in one molecule and
an H attached to an N O or F atom in another molecule see what is a
hydrogen bond CH_3F has an F atom but no H F bonds it

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